



Plant Archives

Journal homepage: <http://www.plantarchives.org>

DOI Url : <https://doi.org/10.51470/PLANTARCHIVES.2026.v26.supplement-2.004>

ZINC DYNAMICS IN SOIL-PLANT SYSTEM AND ITS IMPLICATION FOR CROP NUTRITION- A REVIEW

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(Date of Receiving : 09-02-2026; Date of Revision : 03-03-2026; Date of Acceptance : 18-04-2026)

ABSTRACT

Zinc is an essential micronutrient required for optimum plant growth, metabolic functioning and human nutrition. Zinc deficiency has become widespread in soils particularly in intensively cultivated areas, due to imbalanced fertilizers use, low organic matter, and soil degradation. Role of zinc in plants and its interactions with other nutrients, factors affecting and its availability in soils or its importance for human health. Zinc plays a key role in enzyme activation, protein synthesis, auxin metabolism, membrane integrity, and stress tolerance in plants. Its availability in soil is influenced by factors such as soil pH, texture, organic matter, parent material, and climatic conditions. Nutrient interactions significantly affect zinc dynamics, where it shows synergistic effects with nitrogen and potassium, but antagonistic relationships with phosphorus, calcium, copper, and boron. Zinc deficiency in crops leads to reduced yield and nutritional quality, contributing to human malnutrition, commonly to as hidden hunger. Agronomic biofortification, balanced fertilization, and microbial approaches are effective strategies to enhance zinc availability and uptake. Zinc management through integrated nutrient practices for improving crop productivity, soil health, and nutritional security under changing climatic conditions.

Keywords: Zinc Dynamics, Soil-Plant System, Crop Nutrition.

Introduction

Plants require various nutrients for their growth and development. These are imperative for both agricultural productivity and maintenance soil system for long-term viability which helps to maintain the ecosystem sustainability (Shrivastav *et al.*, 2020). They are important for appropriate biochemical procedures, for improving crop protection against pests and diseases (Yadav *et al.*, 2024). For completing the plant life cycle Arnon and Stout proposed the criteria of essentiality in 1939 to determine whether elements were essential for plant growth and development. An element is considered essential if it satisfies three criteria *viz.* necessary for life cycle completion, direct role in metabolism and nonreplicable (Sharma *et al.*, 2025 A). In agriculture, it has been identified that there are 17 essential nutrients that are necessary for completion of life cycle in plants (Pandey, 2018). Out of these, 3 are non-mineral nutrients (C, H and O) which are taken up from air and water, 6 are macro

nutrients (N, P, K, Ca, Mg and S) which are required in large quantities while 8 are micronutrients (Fe, B, Zn, Ni, Cl, Cu, Mo, Mn) which are required in minute quantities (Kumar *et al.*, 2021). Although, very less amount of micronutrients is required by the plants but they are crucial for optimum growth and development (Gupta *et al.*, 2008). Different elements have different functions in the plant body (Khare *et al.*, 2025). Each nutrient acts as an individual worker that maintains everything functioning properly in plants, which are like individual factories (Bashir *et al.*, 2025). One of the most important components is nitrogen, which helps in the synthesis of proteins, coenzymes, chlorophyll, and even nucleic acids, which are the genetic material of plants (Yang *et al.*, 2025). As the plant's energy manager, phosphorus promotes rapid root development, blooming, and fruiting by promoting energy transmission through ATP (Rather *et al.*, 2025). Potassium functions as a supervisor, controlling vital processes including protein synthesis and flow of

carbohydrates (Mugisha, 2025). By protecting cell membranes and controlling what goes into and out of each cell, calcium helps in maintaining the structural integrity of plants (Adekiya *et al.*, 2025). While micronutrients directly or indirectly affect the vital processes in plants such as respiration, photosynthesis, transpiration lateral growth and reproduction phase (Johnson and Mirza, 2020). Iron supports the creation of chlorophyll and facilitates the transmission of electrons through enzymes, which are necessary for photosynthesis (Jomova *et al.*, 2025). Boron ensures that the plant has energy where it needs it by managing the transfer of sugars and supervising the metabolism of carbohydrates (Álvarez-Herrera *et al.*, 2025). While magnesium functions as an enzyme activator and is a component of the chlorophyll molecule itself, sulfur helps in the construction of plant proteins (Sun *et al.*, 2025).

Zinc also belongs to the category of micronutrients and plays vital role in plant growth and development (Channab *et al.*, 2024). Zinc maintains proper functioning of multiple metabolic processes and supports different kinds of enzyme systems by acting as regulator for them (Tufail *et al.*, 2018). It also helps in synthesis of several growth hormones like indole acetic acid (IAA), cytokinin and gibberellins in plants (Minello *et al.*, 2025). Zinc also promotes starch formation, seed maturation and production (Chen *et al.*, 2025 A). It is important and specific physiological roles in all living membrane to maintaining the structural and functional integrity as well as facilitating gene expression (Kaur *et al.*, 2025). Zinc value is depending on chemical composition of raw materials in the soil. Zinc value of earth's crust is 80 mg/kg approximately and value in soil is usually 10 to 300 mg/kg with an average of 50 mg/kg (Lukaszek-Chmielewska *et al.*, 2025). It is present in different forms in earth's crust, like sulfate, silica and carbonate minerals. Zinc deficiency in plants is frequently observed because low availability of the mineral (Pandey and Gautam 2025). Before Green Revolution soils were naturally fertile as organic manures were used for supplementing depleted soil nutrients however, after green revolution, soil fertility drastically depleted owing to higher cropping intensity and use of synthetic fertilizers which only supplied macro nutrients to the soil resulting in deficiency of micronutrients in the soil (Bashi *et al.*, 2025). On that particular time soils were losing vital nutrients over time that leads to nutrient imbalance and soils were generally nutrient deficient with low nitrogen, phosphorus and depleting in potassium thus resulting in micronutrient deficiencies in plants and ultimately in humans as well as (Das *et al.*, 2022). After Green

Revolution with high-yielding varieties, chemical fertilizers and irrigation initially improved soil nutrient status (Meetei *et al.*, 2025) but long term impacts included nutrient imbalance (high N, high P, low K), widespread micronutrient deficiencies (Zn, Fe, S), declining organic matter, and increasing soil degradation due to input intensive farming (Dimkpa *et al.*, 2023). Micronutrient deficiency occurs when essential plant nutrients required in small amounts are not present in sufficient levels in the soil and that are unavailable for plant uptake (Saleem *et al.*, 2023). This induced the poor plant growth, reduced yield, and characteristic deficiency symptoms. Main causes of micronutrient deficiencies are imbalanced fertilizer use (high N and P, low micronutrients), low organic matter in soil, continuous cropping without replenishment that transform the pH of soil (Radulov and Berbecea, 2024). High soil pH reduces Zn, Fe, Mn availability, soil erosion and leaching (Regasa *et al.*, 2024). Micronutrient deficiency in plants occurs when soil does not provide sufficient amount of essential micronutrients required for normal plant growth and metabolism (Lilay *et al.*, 2024). Even plants need micronutrients in very small quantities, their deficiency causes physiological disorders in plants like stunted growth, curling of leaves, lower flowering and fruit setting, etc (Morya *et al.*, 2023). The most sensitive plants to zinc deficiency include beans, maize, onions, sorghum, and rice; barley, potatoes, soybeans, sugar beet, and tomatoes are moderately sensitive, and forage plants are resistant (Jattan *et al.*, 2023).

According to WHO report 2024-25 micronutrient deficiency also known as hidden hunger and remains a major global health problem. Over 2 billion people worldwide are estimated to be deficient in one or more key micronutrients (Koshyb 2024). Deficiencies of iron, iodine, vitamin A, zinc and other vitamins, minerals are very common. Most of populations especially children and women are affected (Kiani *et al.*, 2022). According to WHO estimate 30.7% of women aged under 15–49 are anemic and among of them 35.5% are pregnant women (Vijayakumar *et al.*, 2025). Zinc deficiency in humans estimated that 15–17 % of the global population (1.1 billion people) is at risk of zinc undernourishment that causes a wide range of human problem like hair losses, weak immune system, impaired wound healing, hypogonadism in males, neurological issues and much more (Hussain *et al.*, 2022). Zinc deficiency is considered as public-health problem in developing countries, especially in South Asia and sub-Saharan Africa (Gupta *et al.*, 2020). In India around 17–18% children of 1–9 years age and about 31% teenager of 10–19 years age have zinc deficiency according to the National Nutrition Survey

(Venkatesh *et al.*, 2021). National nutrition data suggest that overall 25% of population may have inadequate dietary zinc intake in India (Kumar *et al.*, 2025).

To overcome this limitation agronomic biofortification can be a very effective strategy (Matumba *et al.*, 2025). This strategy is particularly important to overcome problems caused by hidden hunger that affects primarily women and young children, often due to an insufficient intake of micronutrients specially zinc (Yilmaz and Yilmaz 2025). This technique aims to increase the bioavailability of nutrient content in parts of plants such as roots, leaves, and grains to improve the dietary quality and public health (Singh *et al.*, 2016). Zinc also exhibits antagonistic interactions with copper and phosphorus which significantly influences the uptake bioavailability and quantitative status of nutrients. Due to this reason the interaction between Zinc, phosphorus and copper supplying of zinc as foliar spray is of great importance (Gong *et al.*, 2020). Deficiency of zinc is increasing day by day and if proper management techniques are not implemented then security is at uptake especially in developing and underdeveloped countries (Singh and Maurya 2025).

Role of zinc

Role of zinc in plants

Zinc is a crucial micronutrient for crop growth. It plays a part in the synthesis of tryptophan, a precursor

to indole acetic acid (IAA), and the hydrogenation of carbonic anhydrates (Kaur *et al.*, 2025). Plant enzymes involved in protein synthesis, auxin synthesis, pollen development, cellular membrane integrity, and carbohydrate metabolism have been stimulated by zinc (Shao *et al.*, 2023). It also regulates cytochrome synthesis and ribosomal fragments. Aldolase, particularly assists in the cellular breakdown of carbon, gets boosted by zinc (Kaur *et al.*, 2025). Applying zinc has enhanced crop quality and productivity (Ashikuzzaman *et al.*, 2024). Higher zinc concentrations can be noxious to plants, inhibiting cell division and elongation and reducing biomass production (Qian *et al.*, 2023). An adequate quantity of zinc may minimize the damaging effects of temporary heat and salt stress and also enhance drought tolerance in several kinds of crops, such as wheat, sunflower, tomato, and red cabbage (Ahmed *et al.*, 2023). The gene expression necessary for plants to withstand stressful conditions is regulated and maintained by zinc (Moulick *et al.*, 2023). Plants that have been deficient in zinc suffer from abnormalities which display deficiency symptoms such as stunted growth, chlorosis, smaller leaves, and plants may become susceptible to damage from intense light, extreme temperatures, and fungal diseases (Ghimirey *et al.*, 2024). Zinc affects a plant's ability to absorb and transport water (Singh *et al.*, 2023).

Table 1: Role of Zinc in Plants

S. No	Role of zinc	Mechanism	References
1	Development of pollen tube	Help full in fertilization of pollen grain	Huang <i>et al.</i> , 2026
2	Resistance to diseases	Adequate zinc improves plant tolerance to disease and environment strees	Wang <i>et al.</i> , 2026
3	Ribosomal fraction stabilizes	Key factor in structural integrity	Tsonev and Cebola, 2012
4	Tranlocation and transport	Zinc enhance the development of vascular tissues (Xylum and Phloem)	Hellmann <i>et al.</i> , 2018
5	Auxin metabolism	Crucial plant hormone regulates growth and development	Hamzah <i>et al.</i> , 2022
6	Protein synthesis	Formation of proteins which are essential for plant growth and development (Dehydrogenase Enzymes, Peptidases, aminoacyl-tRNA synthetase)	Sharma <i>et al.</i> , 2018
7	Synthesis of Cytocrome-C	Important role in respiration and energy production (ATP)	Sen <i>et al.</i> , 2026
8	Nitrogen metabolism	It helps in nitrogen absorption and its conversion into amino acids in plants.	Ashwini <i>et al.</i> , 2026
9	Membrane integrity	Maintain cell membrane stability and protect cells from damage	Tufail <i>et al.</i> , 2018
10	Carbohydrates metabolism	It helps in starch formation and regulates sugar conversion and energy supply.	Suganya <i>et al.</i> , 2020

Role of zinc in humans

Even humans and animals are susceptible to zinc deficiency (Kaur *et al.*, 2025). Zinc deficiency affects 30% of the world's population (Borpatra *et al.*, 2025). In the process of helping to strengthen the human immune system, zinc plays essential roles in cellular processes in living beings (Jomova *et al.*, 2025). Adult humans should consume 8-11 milligrams of zinc per day. In several bodily enzymes, zinc serves as a structural or catalytic component (Temnik *et al.*, 2025). Zinc insufficient levels may result from improper zinc absorption in the body. Human beings will suffer from memory loss, hair loss, problems with the skin, and muscle weakness as a result of a zinc shortage (Patil *et al.*, 2025). A fetus's brain development may be affected by inadequate zinc consumption during pregnancy. Men who are deficient in zinc may experience infertility (Berger *et al.*, 2025). The average daily need for zinc is 11 mg for men and 9 mg for women. Women require 13–14 mg of zinc per day throughout pregnancy and lactation (Zegeye *et al.*, 2025). Children need 3 mg of zinc per day between the ages of 7 months and 3-5 mg between the ages of 4-8, and 8 mg between the ages of 9 and 13. Pig, poultry, breakfast cereals, nuts including roasted peanuts, almonds, walnuts, and oats, and dairy products like yogurt, cheese, and milk are among the foods high in zinc (Monroy *et al.*, 2025).

Zinc deficiencies in Indian soils

Zinc levels in India vary depending on the kind and quality of the soil (Ramalakshmi *et al.*, 2025). Variations in parent material types, soil formation processes, climate, and many other factors contribute to the variation (Haensel 2025). Zinc is abundant in soils formed from igneous rocks and replaces similar-sized ions in silicates, such as Mg^{+2} and Fe^{+2} . Compared to limestone, shale, and igneous rocks, sandstone-derived soils have a lower zinc content (Ghosh *et al.*, 2025). Granite derived soils were found to contain more available zinc than gneiss and basalt (Akingboye *et al.*, 2025). Laterite soils in Tamil Nadu have the highest accessible zinc content, followed by red, alluvial, and black soils (Sarkar and Das 2025). Low total zinc content, high pH, high calcite and organic matter content, and high concentrations of sodium, calcium, magnesium, bicarbonate, and phosphate in the soil solution or labile forms are the primary soil variables that influence zinc availability to plants (Cyprichová *et al.*, 2025). Intensively farmed soils, improperly drained soils, and peat soils frequently have zinc deficiencies (Kadyan and Kumari 2025). Course texture soils have negative interaction with zinc because it has leaching losses, low cation

exchange capacity, low organic matter and poor nutrients retentions while heavily weathered acid soils have high zinc availability due to low pH, but zinc availability reduced by fixing with iron and aluminum oxides present in soils and high accessible phosphorus and silicon has negative interaction with zinc because of nutrient antagonism it makes zinc insoluble and immobile, with all these reasons zinc availability reduce to plants. (Kahsu *et al.*, 2025). Coarse texture soils with low to medium organic carbon content in area like Rajasthan are generally deficient in zinc (Rana and Sharma, 2025). After phosphorus and nitrogen, zinc is regarded as the third major nutrient that limits crop production in India (Makdoh *et al.*, 2025). Due to ongoing soil fertility depletion, zinc deficiency in Indian soils has risen from 42% in 1970 to 63% in 2025 (Behera *et al.*, 2025).

Calcareous soils (pH>7) with high bicarbonate (HCO_3^-) in the soil solution and moderate to high organic matter content (>1.5% organic C) are probably zinc deficient (Sharma *et al.*, 2025 B). Zinc deficiency may also be indicated by a soil exchangeable Mg:Ca ratio greater than 1 (Hosseiny and Maftoun 2025). Due to the production of insoluble zinc hydroxide and related carbonates, 50% of India's soils are mostly zinc deficient (Mued *et al.*, 2025). Zinc is quickly precipitated after being absorbed into the soil colloids and producing chelates. As a result, plants access to zinc rapidly declines (Zhang *et al.*, 2025 A). The crop's ability to absorb zinc may be impacted by high phosphorus application in calcareous soil (Morsy *et al.*, 2025). Seven distinct pools of zinc can be found in soils: carbonate and amorphous oxide bound zinc, organically bound zinc, complexed zinc, water soluble + exchangeable zinc, and zinc pool and zinc bonded by crystalline oxide (Korai *et al.*, 2025). The percentage and concentration of each zinc fraction to total zinc in soil changes the behavior of zinc in soils and its availability of zinc to plants (Umair *et al.*, 2020).

Factors affecting availability of zinc in soil

Numerous factors, including parent material, soil pH, organic matter, texture, calcium carbonate content, phosphate fertilizer application, soil flooding, interaction with other nutrients, temperature etc affect the availability of zinc in the soil (Kumari *et al.*, 2025).

Parent material

Parent material is the primary factor influencing the zinc concentration in the soils (Mu *et al.*, 2025). Zinc content varies among rocks and sediments. Mafic igneous rocks, basalt, and shale have comparatively high zinc contents (Beaudry *et al.*, 2025). Zinc content is often low in quartz-rich materials, granite, and

sandstone. Total and accessible zinc levels are generally higher in soils generated from parent materials rich in zinc (Kumar *et al.*, 2025 A). Zinc is more easily released into the soil solution by parent materials that weather quickly, such as basalt or volcanic ash (Sharma *et al.*, 2025 C). Zinc is released very slowly from slow-weathering rocks (such as quartz or sandstones), which lowers its availability (Hilgard 2025). Compared to illite or bentonite, kaolinite fixes less zinc. As a result, clays with greater CECs, such as bentonite and illite, help fix zinc more firmly, preventing plants from using it (Sayahi *et al.*, 2025).

Soil texture

Sand and other lighter-textured soils have low zinc content (Markoski *et al.*, 2025). Because they have higher CEC values, soils with finer textures like clay have more reactive sites and can hold into more zinc than soils with lighter textures (Yousuf *et al.*, 2025). Zinc adsorption capability is higher in heavier textured soils with greater CEC than in lighter textured soils (Ghosh *et al.*, 2025). Consequently, sandy soils are more prone to zinc shortage than clayey soils. Zinc is adsorbed by clay soils, and pH and CEC regulate this process (Isaboke *et al.*, 2025).

Soil pH

Zinc availability rises when the pH of the soil decreases. The majority of zinc-deficient soils are calcareous and neutral (Athar *et al.*, 2025). Zinc forms the insoluble amorphous ZnFe_2O_4 and ZnSiO_4 precipitates at high pH levels (Muensita *et al.*, 2025). Zn^{2+} uptake decrease in acidic soils that are particularly deficient in zinc due to pH impact (Wang *et al.*, 2025). Dolomite absorbs zinc less readily than magnesite (Leybourne *et al.*, 2025). In calcareous soils, zinc is absorbed by CaCO_3 , which is partially responsible for Zn^{2+} availability (Wali *et al.* 2025); however, when soil calcium carbonate levels rise, zinc availability falls (Sharma *et al.*, 2025 C). The zinc solution in the soil is additionally reduced by iron and aluminum oxides, and this reduction increases as pH rises (Regasa *et al.*, 2025).

Climatic conditions

Zinc availability rises in warm, dry climates, while zinc insufficiency is more prevalent in chilly, rainy seasons (Nayak *et al.*, 2025). Low temperatures, high moisture content, and low light levels can all lead to a zinc deficit (Poudel *et al.*, 2025). Conversely, higher soil temperatures improve zinc availability by increasing Zn^{2+} solubility and diffusion (Ahmed *et al.*, 2025). Flooded soils are more frequently linked to zinc deficiency than dry soils. Flooding-related zinc

shortage was caused by a zinc interaction with free sulfide (Jadhav *et al.*, 2025). Zinc transforms into amorphous sesquioxide under submerged circumstances, precipitating ZnFe_2O_4 decrease the rate of zinc absorption by plant roots (Dhaliwala *et al.*, 2025).

Soil organic matter

Zinc forms both soluble and insoluble complexes with high molecular weight organic compounds, such as fulvic and humic acids (Boguta and Sokołowska, 2020). When humic soils and insoluble peats are combined, zinc availability decreases (Yasir *et al.*, 2025). Zinc forms a chelated zinc complex in the mineral soil, increasing the soil's zinc solution. Zinc in soil can be chelated by recently applied organic material (Sethi *et al.*, 2025).

Interaction with nutrients

Nitrogen and zinc shows positive response with each other. Nitrogen deficient soils which contain all other essential nutrients do not response to zinc application and plants became zinc deficient (Bazie *et al.*, 2025).

The zinc shortage prevalent in soils is caused by high phosphorus availability (Rehman *et al.*, 2025). Zinc concentration in shoots of plants decreases by effect of growth (dilution effect) which signifies that zinc concentration in plant decrease by increasing plant growth and its concentration decreases in plant tissues. Plants with low zinc increase the phosphorus concentration, which results in hazardous symptoms that resemble zinc deficiencies (Singh *et al.*, 2025). Whereas mycorrhizae can have increased phosphorus intake and result in zinc shortage (Bhardwaj *et al.*, 2025).

High Calcium content inhibits the absorption of zinc by plants from soil. Zinc concentrations positively decreased with increasing Calcium concentrations in soil solutions (Chatzissavvidis *et al.*, 2025). They found that increasing of calcium from 0 mM to 40 mM inhibited the rate of zinc absorption. However higher calcium (100 mM) had no additional effect on zinc absorption (Orth *et al.*, 2025).

Sulphate and chloride are water soluble and it cannot create any problem in the availability of zinc to plants and while borate and molybdate are present in very small amounts so it is not a problem to the zinc availability to plants (Meyvisch *et al.*, 2025).

Nutrients dynamics and their interaction with zinc

Plant growth and development are largely determined by macronutrients and micronutrient availability. To ensure beter productivity of crops plants, it is essential to understand the plants nutrients

uptake, transport, assimilation and their biological interactions (Kumar *et al.*, 2021). An influence or reciprocal action of one element upon another in relation to plant growth is one definition of interaction (Stein *et al.*, 2025). The way plants react differently to one ingredient combined with varied concentrations of a second element applied concurrently is another factor (Hashempur *et al.*, 2025). These kinds of interactions can occur both inside the plant and in the soil. When giving plants a sufficient quantity of micronutrients, these interactions should be considered (Richards *et al.*, 2025). Zinc's availability from soils and its status in the plant during the growth process particularly its absorption, distribution, or utilization can be influenced by nutrients (Usmani *et al.*, 2025).

Zinc interaction with macronutrient

Zinc-nitrogen interaction

Nitrogen fertilizers can help plants that are deficient in zinc. Since the administration of nitrogen encourages plant growth, it is possible to provide favorable interactions between growing levels of zinc and nitrogen fertilizers (Virkhare *et al.*, 2025). The outcome of nitrogen application serves to promote plant growth and, to a lesser extent, changes the pH of the root environment (Das *et al.*, 2025). Excessive application of nitrogen fertilizers in the zinc deficient soils do not improve zinc uptake in root, shoot and grain (Xue *et al.*, 2021). Because ammonium ions have an acidifying impact on soil, application of zinc sulphate (ZnSO_4) in conjunction with nitrogen fertilizers effectively control zinc shortage, whereas ZnSO_4 alone had little effect (Almendros *et al.*, 2025).

Zinc-phosphorus interaction

Since 1936, research on the relationship between zinc and phosphorus has been conducted. Research on this significant plant growth problem is ongoing (Kumar *et al.*, 2025 B). Phosphorus induced zinc deficiency is the common term used to describe the interaction (Zhang *et al.*, 2025 A). Phosphorus application to the soil or high levels of accessible phosphorus is linked to this problem in plant growth (Jahan *et al.*, 2025). The use of zinc fertilizers helps stop the symptoms of zinc insufficiency (Sethi *et al.*, 2025). Phosphorus-induced zinc shortage has been linked to four general causes (Miruna *et al.*, 2025). These include

- i. The interaction between phosphorus and zinc in the soil
- ii. A slower rate of zinc translocation from the roots to the shoots

- iii. A straight forward dilution effect on zinc concentration in plant tops as a result of growth responses to phosphorus
- iv. A metabolic disorder within plant cells associated with an imbalance between phosphorus and zinc

Lowering of zinc concentration to inadequate levels in soil due to the production of an insoluble zinc phosphate [$\text{Zn}_3(\text{PO}_4)_2$] (Sazanova *et al.*, 2025). Applied phosphorus has been shown by numerous researchers to elicit signs of zinc shortage in plants (Khan *et al.* 2023). Both the total zinc contents and the zinc concentrations in the plant tops were lowered by the increased phosphorus levels in the soil (Abbas *et al.*, 2025). Zinc and phosphorus have an antagonistic effect on plant roots. The antagonistic effect of phosphorus on zinc can be lessened by a balanced fertilizer ratio, and the total zinc content can either rise or stay the same (Shahzad *et al.*, 2025). Phosphorus interferes with the absorption, transport, and use of zinc, which is the primary cause of phosphorus-induced zinc insufficiency (Samal *et al.* 2025).

Zinc-potassium Interaction

Zinc and potassium interaction in the soil plant system is generally synergistic in nature (Pavithra *et al.*, 2025). Adequate potassium nutrition improves root growth, enzyme activation, and membrane permeability which enhances the uptake and translocation of zinc within the plant (Karthika *et al.*, 2018). Potassium also helps in efficient utilization of absorbed zinc by reducing the zinc deficiency symptoms (Hacisalihoglu 2020). Zinc support better potassium uptake by maintaining membrane integrity and normal metabolic activity (Wang *et al.*, 2013). There is no direct chemical antagonism between zinc and potassium in soil and also they not form any insoluble complexes (Bashir *et al.*, 2025). However excessive potassium application under certain conditions may indirectly reduce zinc availability by slightly increasing soil pH especially in calcareous soils (Demirkaya and Gulser, 2025). In general, balanced application of zinc and potassium fertilizers promotes better nutrient use efficiency and crop productivity (Sebhatleab *et al.*, 2025).

Zinc-calcium Interaction

Plants cannot readily access calcium zincate [$\text{CaZn}_2(\text{OH})_6 \cdot 2\text{H}_2\text{O}$], which is formed when calcium carbonate reacts with zinc in soil solution (Nadim and Khitab, 2023). $K_{sp} = 43.9$ is the solubility product for calcium zincate which is a large value that indicate compound dissolve in alkaline condition (Chaudhary, 2023). Calcium zincate after dissolve in alkaline condition [$\text{CaZn}_2(\text{OH})_6 \rightleftharpoons \text{Ca}^{2+} + 2\text{Zn}^{2+} + 6\text{OH}^-$] releases

Ca^{2+} and Zn^{2+} ions and act as temporary zinc source that is available to plants (Feng *et al.*, 2023). Calcium zincates ionic strength is 0.1 M in a pH range between 11.4 to 12.7 (Rashid, 2020). The availability of calcium decreases by high zinc application. Zinc and calcium have long been known to have antagonistic effects (Haleema *et al.*, 2024). Zinc absorption and translocation in roots was both reduced when calcium content in soil solution increased (Chahal *et al.*, 2023). In certain instances, zinc fertilization reduced the amount of calcium in leaves (Pimentel *et al.*, 2023). The application of calcium helps in mitigating zinc toxicity. Under conditions of zinc toxicity, calcium stabilizes the plasma membrane and plays a significant role in cell permeability (Costa *et al.*, 2023). The crops cultivated in limestone soils had lower zinc concentrations in plant biomass than those grown in acidic soils and metal-contaminated soils (Lahori *et al.*, 2020).

Zinc-magnesium interaction

There is positive interaction between zinc and magnesium (Hamedifard *et al.*, 2020). The concentration of zinc in crop biomass increased on magnesium application and vice-versa (Sadeghi *et al.*, 2021). By choosing the lime materials like CaCO_3 , $\text{CaMg}(\text{CO}_3)_2$, CaO , CaOH_2 based on their magnesium content present in the soil, zinc availability reduced to plants uptake (Minato *et al.*, 2023). Using CaCO_3 and MgCO_3 in a 1:1 or 2:1 ratio produced the best plant development, while using a 5:1 ratio resulted in less growth (Umer *et al.*, 2020).

Zinc-sulphur interaction

Zinc and sulphur have both antagonistic and complementary effects (Hemasree *et al.*, 2025). Sulphur application reduced the zinc concentration in shoots and inhibited the production of dry matter production (Vidal *et al.*, 2025). An application of 100 kg S/ha resulted in a modest decrease in the content of zinc in *kharif* crop plants (Magodia *et al.*, 2025). However, the concentration of zinc in *rabi* crop plants increased significantly when sulphur was applied (Rais, *et al.*, 2025). Zinc and sulphur application are positive correlated with growth characteristics, yield attributes, and yield of various crop plants with their synergistic effect and sulphur helps in zinc uptake, protein synthesis, enzyme activation etc (Ghanem *et al.*, 2025).

Zinc interaction with micronutrient

Zinc-copper interaction

Zinc significantly reduces the absorption of copper. Copper alters the distribution of zinc in plants

by preventing zinc absorption (Rolic *et al.*, 2025). Crops grown on copper and zinc-deficient soils have been shown to have a very strong copper-zinc antagonistic relationship (Gao *et al.*, 2025). Zinc significantly reduces copper uptake in plants, due to competitive suppression of zinc on copper absorption (Ouyang *et al.*, 2025). Further it has also been proven that copper ions competitively impede zinc absorption (Chen *et al.*, 2025 (A) Ullah *et al.*, 2025). Copper and zinc ions act different in soil and soil solution owing to adsorption, precipitation and buffering processes in soil that are absent in soil solution (Solymos *et al.*, 2025). Zinc and copper existed as divalent ions in solution, but in majority of soils, they are mostly found in complex forms, with a significantly higher percentage of complexed copper than zinc (Rocco *et al.*, 2025). Zn^{2+} activity would therefore be significantly higher than Cu^{2+} activity at the absorbing sites, making it an efficient competitor against absorption of copper (Gautam *et al.*, 2025).

Zinc-iron interaction

Similar to the connection between zinc and phosphorus, the interaction between zinc and iron is equally complex (Sethi *et al.*, 2025). The content of zinc in plant tissues decreases with increasing iron levels (Ozkutlu and Korkmaz, 2025). It has been demonstrated that an increase in soil iron concentration slows down the pace at which plant roots absorb zinc (Kumar *et al.*, 2025 C). However, the rate at which the crops absorbed zinc remain unaffected by iron at low concentration (10 μM) (Suganya *et al.*, 2020). Higher concentration of iron (100 μM Fe^{2+}) found in flooded soils totally inhibited crops' ability to absorb zinc (Deng *et al.*, 2009). Zinc concentrations in plant shoots and the rate of zinc absorption in grasses and dicotyledonous plants were both elevated by iron deprivation (Mateos Naranjo *et al.*, 2008). Zinc absorption in dicotyledonous plants can be enhanced by the acidification of the rhizosphere in iron deficiency soils (Molnar, *et al.*, 2023), while in grasses, the release of phytosiderophores causes higher rate of zinc absorption also to improved mobilization of zinc from calcareous soils (Kumar *et al.*, 2025 D). Similarly, acidification of the rhizosphere and the release of reductants and phytosiderophores may be the cause of iron accumulation in the shoots of zinc-deficient plants (Leskova *et al.*, 2022).

Zinc-boron interaction

There is an antagonistic link between the buildup of zinc and boron in plants (Long and Peng, 2023). Under zinc deficiency, plants will accumulate more boron, but applying zinc fertilizers will reduce it

(Shahgholi *et al.*, 2023). Zinc fertilizers can successfully reduce boron toxicity to crops in soil with excessive boron concentrations (Zhao *et al.*, 2024). However, applying boron fertilizers to crops has not been successful in lowering zinc buildup (Okasha *et al.*, 2024). Zinc and boron can show synergistic interaction under extreme conditions of boron deficiency or excess boron. Excess boron can enhance zinc uptake and translocation which may result in zinc toxicity symptoms (Gulut, 2024).

Zinc toxicity

Excessive zinc concentrations can be hazardous to plants. Zinc poisoning reduced the growth and development of leaves and roots (Wei *et al.*, 2022). As zinc concentration rises, plant chloroplasts produce less NADPH, and RUBP carboxylase enzyme (Chen *et al.*, 2022). Zinc toxicity lowers ATP synthesis, which in turn decreases chloroplast activity and photosynthesis (Sperdoui *et al.*, 2022). Iron and phosphorus uptake is decreased by high zinc levels. Zinc toxicity is caused at more than 300 ppm of zinc in plants tissue (Angon *et al.*, 2024).

Future prospects

Due to increased incidence of macronutrients and micronutrients deficiencies in humans there is a need to improve the nutritional value in humans diet (Avnee *et al.*, 2023). There are several bio-fortification strategies introduced like traditional breeding, transgenic, agronomic, and modernized agriculture practices which are employed to biofortify crops that fulfill the nutritional needs (Jaiswal *et al.*, 2022). However, these strategies are also lacking sustainability due to the specific crop species and micronutrients provision of long-term monitoring and assistance (Jaiswal *et al.*, 2022). High cost of special chemical fertilizers induced alteration in soil metabolism, environment flow and agriculture sustainability (Saikanth *et al.*, 2023) remain major challenge for nutrient management. Therefore, microbial biofortification is imperative and should be emphasized for exploring the plant microbe interaction within specific soil system and climatic condition. Plant uptake of micronutrient (Fe, Zn, Mg, Ca, Se, and S) and there availability in the soil system can be efficiently enhanced through such microbe based approaches (Chung *et al.*, 2022). At present microbial biofortification provides a sustainable strategy for improving macro and micro nutrients of crops through balanced soil system, ultimately contributing to climate resilient agriculture (Jaiswal *et al.*, 2022).

Conclusion

Zinc is a vital micronutrient that plays a central role in plant growth, metabolism, stress tolerance, and human nutrition. Intensive agriculture, imbalanced fertilizer use and soil degradation have lead to widespread zinc deficiency in soils, crops and ultimately human diets particularly in developing countries. Although at the regional level, a thorough 4R (right site, right time, right source, and right dose) strategy to zinc application can be created for various crops, and the ideal combination can be discovered to achieve high grain zinc concentration and also studied nutrient interactions, especially with nitrogen, phosphorus, calcium, copper, iron, and boron, strongly influence zinc availability and uptake in soil plant systems. Understanding these interactions is essential for improving zinc use efficiency and crop productivity. Continuous zinc administration at the basal dose may be able to resolve the conflict with zinc deficiency in soil conditions. However, under stressful situations, agronomic biofortification might be an alternative, and its impacts must be assessed. More weather variations are anticipated as a result of climate change, necessitating the development of a stress-resistant biofortification strategy. When compared to other application techniques, foliar spraying yields the highest bioavailability of zinc.

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